Safety Assessment of Hydrofluorocarbon 152a as Used in Cosmetics

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Abstract

The Expert Panel for Cosmetic Ingredient Safety (Panel) reviewed the safety of Hydrofluorocarbon 152a, which functions as a propellant in personal care products. The Panel reviewed relevant data provided in this safety assessment, and concluded that Hydrofluorocarbon 152a is safe in the present practices of use and concentration described in this safety assessment.

Keywords

hydrofluorocarbon 152a, safety, cosmetics

Introduction

Hydrofluorocarbon 152a is a gas that functions as a propellant in personal care products, according to the *International Cosmetic Dictionary and Handbook*.¹ It is commonly known as 1,1-difluoroethane.

Chemistry

Definition

Hydrofluorocarbon 152a (CAS No. 75-37-6) is the halocarbon, 1,1-difluoroethane, that conforms to the formula CH_3CHF_2 (Figure 1).¹

Chemical and Physical Properties

Hydrofluorocarbon 152a is a colorless, odorless gas, with a vapor pressure of 4550 mmHg at 25°C. This ingredient is distinct from chlorofluorocarbon propellants, such as hydro-chlorofluorocarbon 142b, because there are no chlorine atoms to react with stratospheric ozone. Additional physical and chemical properties of Hydrofluorocarbon 152a are provided in Table 1.

Method of Manufacturing

Hydrofluorocarbon 152a may be derived by reacting hydrogen fluoride with acetylene.² The material may also be produced by a catalytic reaction of vinyl chloride with hydrofluoric acid in a closed system.3

Impurities

Hydrofluorocarbon 152a is reported to be greater than 99.9% pure.³ Impurities may include water, residual hydrochloric acid, and residual hydrofluoric acid.

Use

Cosmetic

The safety of the cosmetic ingredient included in this assessment is evaluated based on data received from the U.S. Food and Drug Administration (FDA) and the cosmetics industry on the expected use of this ingredient in cosmetics. Use frequencies of individual ingredients in cosmetics are collected from manufacturers and reported by cosmetic product category in FDA's Voluntary Cosmetic Registration Program (VCRP) database. Use concentration data are submitted by Industry in response to surveys, conducted by the Personal Care Products Council (Council), of maximum reported use concentrations by product category.

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Bart Heldreth, Executive Director, Cosmetic Ingredient Review, 1620 L Street, NW, Suite 1200, Washington, DC 20036, USA. Email: cirinfo@cir-safety.org Figure 1. Hydrofluorocarbon 152a.

Table 1. Physical and Chemical Properties of Hydrofluorocarbon152a.

Property	Value	Reference
Physical form	Gas	2
Color	Colorless	2
Odor	Odorless	2
Molecular weight (g/mol)	66. l	3
Density (at -25° C)	1.004	2
Vapor pressure (mmHg at 25°C)	4550	3
Henry's law constant (atm·m ³ /mol)	0.02	3
Melting point (°C)	-117	2
Boiling point (°C)	-24.7	2
Water solubility (g/l at 25°C)	2.671 (estimated)	3
Log K _{ow}	0.75	3

According to 2017 VCRP data, Hydrofluorocarbon 152a is used in 467 formulations; the majority of uses are in leave-on hair care products (Table 2).⁴ The results of the concentration of use survey conducted in 2015 by the Council indicate the highest reported maximum concentration of use to be 80% in hair sprays.⁵

This product is believed to be solely used in spray products, like hair spray and spray deodorants. However, because this ingredient is a gas under all exposure conditions, inhalation is possible for all product types in which it is used.

Based on environmental regulations that include a requirement that those products containing fluorocarbons be labeled, Hydrofluorocarbon 152a is not used in personal hygiene or household products in Europe.⁶⁻⁸

Non-Cosmetic

Hydrofluorocarbon 152a may be used as an aerosol propellant, a foam expansion agent, a refrigerant, and as a catalyst regenerator.³

Toxicokinetics

Absorption, Distribution, Metabolism, and Excretion (ADME)

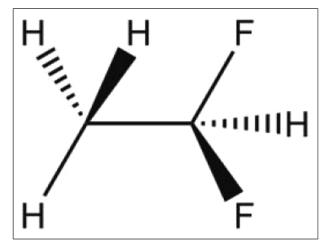
Animal

Inhalation. Metabolites included fluoride ion and a trace of acetyl fluoride in urine collected for nuclear magnetic resonance (NMR) spectroscopy analysis from male CD rats exposed via inhalation to 3000 ppm Hydrofluorocarbon 152a (see Acute Toxicity for further study details).⁹ No fluoroacetate was detected. The time at which the urine was collected after exposure was not identified in the study report.

Human

Inhalation. The uptake, distribution, and elimination of Hydrofluorocarbon 152a were studied in male and female subjects.¹⁰ 6 women and 4 men completed all exposure sessions (0, 200, and 1000 ppm test material), with 1 additional male exposed to 0 and 200 ppm, another male exposed to only 200 ppm, and another male exposed only to 0 ppm (n for 0 ppm = 12, n for 200 ppm = 12, and n for 1000 ppm = 10). Subjects were exposed for 2h on 3 separate occasions to Hydrofluorocarbon 152a, and exposures were performed during light exercise on computer-controlled ergometer bicycles in an exposure chamber (20 m^3) with a controlled climate. The concentration of the test material in the chamber was checked by gas chromatography (GC) at 5 minute intervals throughout the exposure sessions. Mixed exhaled air was collected once before the exposure, 5 times during the exposure, and 7 times after the exposure. Pulmonary ventilation was recorded with an electric spirometer during every breath sampling period. Venous blood was collected from the brachial vein prior to exposure and at 3 h and 22 h after exposure for analysis of inflammatory markers, while arterialized capillary blood was collected from the subjects' finger tips before, during, and after exposure. Urine was sampled once before exposure and at 2, 4, and 6 h after onset of exposure, as well as twice in the evening and once the following morning. The presence of Hydrofluorocarbon 152a in the blood and urine was analyzed by head-space GC. The urine was analyzed for fluoride with an ion selective electrode and for potential metabolites with NMR.

In the blood, initial increases in Hydrofluorocarbon 152a were fast, and average concentrations of 7.4 μ M (for 200 ppm) and 34.3 μ M (for 1000 ppm) were achieved within a few minutes of exposure. Within 4 h post-exposure, the concentration was less than 1% of the steady-state level. Blood concentrations were below detection limits 22 h post-exposure. The area under the curve (AUC) of the test material in blood was 1042 μ M·min at 200 ppm and 4572 μ M·min at 1000 ppm, which indicated dose-proportional kinetics. No exposure-related effects were observed in inflammatory markers in the blood plasma. Total inhaled Hydrofluorocarbon 152a was approximately 20.6 and 99.6 mmol for the 200 and



Total^a

Duration of use Leave-on Rinse off

Exposure type Eye area

Diluted for (bath) use

Incidental ingestion

Dermal contact

Hair-coloring

Nail

Deodorant (underarm)

Hair-non-coloring

Incidental inhalation –Sprays

Incidental inhalation-Powders

	# of Uses 467 ^b	Max Conc of Use (%)
		2-80
	457	2–80
	10	NR
	NR	NR

Table 2. Frequency and Conce

Note. NR = Not reported.

Mucous membrane Baby products

^aBecause each ingredient may be used in cosmetics with multiple exposure types, the sum of all exposure types may not equal the sum of total uses. ^bIt is likely that all 467 reported uses are in spray products, but type of exposure cannot be confirmed.

NR

NR

386

NR

164

57

276

27

NR

4

NR

^cIn spray deodorants.

^dIncludes use as a hair mousse at 3-3.9%.

1000 ppm exposures, respectively. Post-exposure decreases of the test material in exhaled air and urine were similar to that in blood. The AUCs of the test material in urine were 190 µM·min for 200 ppm and 1271 µM·min for 1000 ppm. After exposure to 200 ppm and 1000 ppm Hydrofluorocarbon 152a, about 0.004% and 0.009%, respectively, of the total amount inhaled was excreted in the urine within 23 h. About 20 µmol excess fluoride (0.013% of inhaled) was excreted in urine following exposure to 1000 ppm test material when compared to the control. This was statistically significantly greater than the amount excreted in the urine of both the control and the 200 ppm exposed subjects (P = 0.008), and urinary fluoride excretion remained significantly elevated in the post hoc test (P < .05). Fluoride excretion rate varied; however, it was statistically significantly higher in the first 2 urine samples after exposure to 1000 ppm test material when compared to the control subjects and subjects exposed to 200 ppm (P = .0004). No fluorine-containing metabolites could be detected in the urine, indicating biotransformation of the test material in humans was very low.¹⁰

Toxicological Studies

Acute Toxicity Studies

Acute oral and inhalation studies are summarized in Table 3.^{3,9,11} In rats, the lowest lethal oral dose (LD_{LO}) reported for Hydrofluorocarbon 152a was greater than 1500 mg/ kg, which was the highest dosage tested.³ However, this study was considered invalid by the test sponsors because an unsuitable test system was used. The LC_{50} reported in a mouse inhalation study was 977 200 ppm. Cardiac arrhythmia was observed in dogs exposed for 5 min to 150000 ppm Hydrofluorocarbon 152a in an inhalation study.

Short-Term Toxicity Studies

Inhalation. In a short-term inhalation study of Hydrofluorocarbon 152a, 10 male ChR-CD rats were exposed to 100 000 ppm of the test material in air 6 h/day, 5 days/week, for 2 weeks (exposure method not reported).^{3,11} Following the final exposure, 5 rats were killed for gross and histopathologic examination while the remaining rats were killed after a 14-d recovery period. Hematological, urine analytical, and biochemical indices were measured in all rats before they were killed. During exposure to the test material, the rats appeared to be anesthetized, which was indicated by sleep and unresponsiveness to sound. No other adverse effects were observed. A slight increase in urinary fluoride was observed following the final exposure.

In another short-term study, 8 albino rats (sex not specified) were exposed to 100 000 ppm Hydrofluorocarbon 152a in air 16 h/d for 2 months (exposure method not reported).^{3,11} At the end of the exposure period, the animals were killed and examined for gross pathological changes. Lung and liver sections were examined microscopically. During the exposure period, no clinical signs of toxicity were observed. Necropsy indicated no adverse changes. Mild diffuse infiltration of small

NR

NR 12-80

NR 12-45

16.5-35°

2-80^d

42.3

NR NR

NR

Table 3. Acute Studies.

Concentration/Dose	Study protocol	Results	Reference
Oral			
200, 300, 450, 670, 1000, or 1500 mg/kg dissolved in corn oil (23 or 46 mg/ml)	was then administered via gavage to 1 male Crl:CD(R)BR rat/dose; test material was kept under pressure in aerosol cans maintained in an ice bath; doses greater than 450 mg/kg	$LD_{Lo} > 1500 \text{ mg/kg}$; no mortalities observed, lethargy observed at 1000 and 1500 mg/kg; high carriage, wet and yellow stained perineum, and diarrhea observed in all rats 1 to 2 days post- dosing. This study was considered invalid by test sponsors due to use of an unsuitable test system	3,11
Inhalation			
896 000 to 1 065 000 ppm	Acute study in mice; exposure for 2 h (no further details provided)	LC ₅₀ = 977 200 ppm; narcotic effects observed at this concentration	3,11
3000 ppm	Study of several fluorinated ethanes, 3–4 CD rats exposed for 4 h in a closed, recirculating chamber; urine collected after exposure and analyzed by using NMR (see Toxicokinetics- ADME section)	No adverse effects observed during or after exposure	9
66 400, 175 200, 319 000, 383 000, or 437 500 ppm	Groups of 6 male ChR-CD(R) rats exposed for 4 h in an exposure chamber	LC _{Lo} = 383 000 ppm; 1/6 rats died at 383 000 ppm; 2/6 rats died at 437 500 ppm; labored breathing, lethargy, and unresponsiveness to sound observed during exposure; no clinical signs observed after exposure and no compound-related changes to gross pathology	3,11
100 000 to 550 000 ppm	Acute study in albino rats; exposure for 30 min (no further details provided)	No postural reflex at 200 000 ppm; no righting reflex at 250 000 ppm; no corneal reflex at 450 000 ppm; acute lung irritation at \geq 400 000 ppm; mortality observed at \geq 500 000 ppm after 10–25 min of exposure	3,11
74 000, 100 000, or 200 000 ppm	Acute study in rats; 2 animals per concentration; whole body exposure for 2 h	Occasional trembling and incoordination during exposure; no mortalities	3,11
	Cardiac sensitization study in 12 male Beagle dogs/group; animals exposed for 5 min via a mask on snout; animals received a 0.008 mg/kg intravenous (i.v.) injection of epinephrine before and after exposure	Cardiac arrhythmia observed in 3/12 dogs at 150 000 ppm; no response at 50 000 ppm	3,11
500 000 ppm	Acute study in 3 dogs; 2 dogs also received atropine and succinylcholine via i.v. injection (no further details provided)	Light surgical anesthesia with no change in EKG	3,11

and large round cells in the lung was observed during microscopic examination, which indicated mild chronic irritation.

Chronic Toxicity Studies

See Carcinogenicity section below.

Developmental and Reproductive Toxicity Studies

The maternal and developmental toxicity of Hydrofluorocarbon 152a was investigated in mated CD rats.^{3,11} Groups of 27 female rats were exposed 6 h/d on gestation days 6–15 via whole body inhalation to 0, 5000, or 50 000 ppm. The dams were exposed to the test material in 1.4 m^3 stainless steel and glass chambers under dynamic airflow conditions. The animals were observed daily for signs of toxicity and weighed periodically throughout the study. The dams were killed on gestation day 21, and organs of the thoracic and abdominal cavities and the fetuses were examined.

No treatment-related clinical signs of toxicity or body weight changes were observed in the dams. No statistically significant differences were observed in the numbers of corpora lutea, implantations, resorptions or live fetuses, fetal weight, or crown-rump length between the treatment groups and the controls. The pregnancy ratios at 0, 5000, and 50 000 ppm were 22/27, 21/27, and 19/27, respectively. No gross pathological abnormalities were observed in ovaries, uterine

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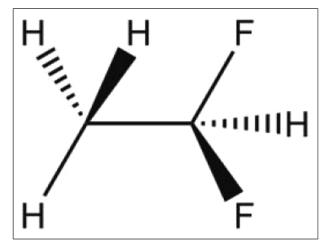
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Total^a

Duration of use Leave-on Rinse off

Exposure type Eye area

Diluted for (bath) use

Incidental ingestion

Dermal contact

Hair-coloring

Nail

Deodorant (underarm)

Hair-non-coloring

Incidental inhalation –Sprays

Incidental inhalation-Powders

	# of Uses 467 ^b	Max Conc of Use (%)
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	457	2–80
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Note. NR = Not reported.

Mucous membrane Baby products

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NR

NR

386

NR

164

57

276

27

NR

4

NR

^cIn spray deodorants.

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In another short-term study, 8 albino rats (sex not specified) were exposed to 100 000 ppm Hydrofluorocarbon 152a in air 16 h/d for 2 months (exposure method not reported).^{3,11} At the end of the exposure period, the animals were killed and examined for gross pathological changes. Lung and liver sections were examined microscopically. During the exposure period, no clinical signs of toxicity were observed. Necropsy indicated no adverse changes. Mild diffuse infiltration of small

NR

NR 12-80

NR 12-45

16.5-35°

2-80^d

42.3

NR NR

NR

Table 3. Acute Studies.

Concentration/Dose	Study protocol	Results	Reference
Oral			
200, 300, 450, 670, 1000, or 1500 mg/kg dissolved in corn oil (23 or 46 mg/ml)	was then administered via gavage to 1 male Crl:CD(R)BR rat/dose; test material was kept under pressure in aerosol cans maintained in an ice bath; doses greater than 450 mg/kg	$LD_{Lo} > 1500 \text{ mg/kg}$; no mortalities observed, lethargy observed at 1000 and 1500 mg/kg; high carriage, wet and yellow stained perineum, and diarrhea observed in all rats 1 to 2 days post- dosing. This study was considered invalid by test sponsors due to use of an unsuitable test system	3,11
Inhalation			
896 000 to 1 065 000 ppm	Acute study in mice; exposure for 2 h (no further details provided)	LC ₅₀ = 977 200 ppm; narcotic effects observed at this concentration	3,11
3000 ppm	Study of several fluorinated ethanes, 3–4 CD rats exposed for 4 h in a closed, recirculating chamber; urine collected after exposure and analyzed by using NMR (see Toxicokinetics- ADME section)	No adverse effects observed during or after exposure	9
66 400, 175 200, 319 000, 383 000, or 437 500 ppm	Groups of 6 male ChR-CD(R) rats exposed for 4 h in an exposure chamber	LC _{Lo} = 383 000 ppm; 1/6 rats died at 383 000 ppm; 2/6 rats died at 437 500 ppm; labored breathing, lethargy, and unresponsiveness to sound observed during exposure; no clinical signs observed after exposure and no compound-related changes to gross pathology	3,11
100 000 to 550 000 ppm	Acute study in albino rats; exposure for 30 min (no further details provided)	No postural reflex at 200 000 ppm; no righting reflex at 250 000 ppm; no corneal reflex at 450 000 ppm; acute lung irritation at \geq 400 000 ppm; mortality observed at \geq 500 000 ppm after 10–25 min of exposure	3,11
74 000, 100 000, or 200 000 ppm	Acute study in rats; 2 animals per concentration; whole body exposure for 2 h	Occasional trembling and incoordination during exposure; no mortalities	3,11
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and large round cells in the lung was observed during microscopic examination, which indicated mild chronic irritation.

Chronic Toxicity Studies

See Carcinogenicity section below.

Developmental and Reproductive Toxicity Studies

The maternal and developmental toxicity of Hydrofluorocarbon 152a was investigated in mated CD rats.^{3,11} Groups of 27 female rats were exposed 6 h/d on gestation days 6–15 via whole body inhalation to 0, 5000, or 50 000 ppm. The dams were exposed to the test material in 1.4 m^3 stainless steel and glass chambers under dynamic airflow conditions. The animals were observed daily for signs of toxicity and weighed periodically throughout the study. The dams were killed on gestation day 21, and organs of the thoracic and abdominal cavities and the fetuses were examined.

No treatment-related clinical signs of toxicity or body weight changes were observed in the dams. No statistically significant differences were observed in the numbers of corpora lutea, implantations, resorptions or live fetuses, fetal weight, or crown-rump length between the treatment groups and the controls. The pregnancy ratios at 0, 5000, and 50 000 ppm were 22/27, 21/27, and 19/27, respectively. No gross pathological abnormalities were observed in ovaries, uterine

horns, vital organs, or tissues of the treated animals. External, skeletal, and internal examinations of fetuses revealed no evidence of teratogenicity. The no observed effect level (NOEL) for maternal and developmental toxicity in rats was 50 000 ppm, which was the highest concentration tested in this study.^{3,11}

Genotoxicity Studies

In vitro and in vivo genotoxicity studies are summarized in Table 4.^{3,11,12} Hydrofluorocarbon 152a was not mutagenic in Ames tests at concentrations up to 75%, but it was weakly clastogenic at concentrations of 60% and 70% (19 h exposure, without metabolic activation) in a chromosomal aberration test in human lymphocytes. Hydrofluorocarbon 152a was not genotoxic in a micronucleus assay in which rats were exposed by inhalation to concentrations up to 19 500 ppm.

Carcinogenicity Studies

In a 2-year inhalation study, male and female Crl:CD(R)Br rats were exposed to 0, 2000, 10000, or 25000 ppm

Hydrofluorocarbon 152a in air 6 h/d, 5 d/wk.^{3,11} There were 30 rats/sex in each exposure group, and rats were 54 d old at the first exposure. Rats were exposed whole body to the test material in chambers. Body weights were recorded twice monthly for the first 14 weeks and then once a month for the rest of the study. Animals were observed for clinical signs of toxicity twice daily during the work week while animals were observed daily for mortality on the weekends and holidays. Ten rats/sex/dose underwent clinical pathology evaluation at 1, 3, 6, 12, 18, and 24 month, which included hematology and clinical chemistry studies. Urine was collected and analyzed the day prior to blood collection. Ten rats/sex/dose were killed and necropsied at 3 and 12 months and all remaining surviving animals were killed and necropsied at 24 mo. Gross examinations were performed on all rats and select tissues underwent microscopic examination. Organ weights were recorded and histopathological examinations were conducted on the control and high-dose groups and on any animals that died during the study. Kidney and nasal tissues at the 3 and 24 months killings were evaluated from all low- and mid-dose groups.

 Table 4. Genotoxicity Studies.

Concentration/Dose	Study protocol	Results	Reference
In vitro			
0%, 20%, 30%, 40%, 50%, or 75%	Bacterial reverse mutation assay (Ames test) in Salmonella typhimurium strains TA97a, TA98, TA100, and TA1535 and Escherichia coli strain WP2uvrA (pKm101), with and without S9 metabolic activation (in accordance with OECD TG 471); plates exposed to test material in glass chambers	Not mutagenic	3,11
0%, 20%, 35%, or 50%	Ames test in S typhimurium strains TA98, TA100, TA1535, and TA 1537 with and without metabolic activation by S9; no further details provided	Not mutagenic	3,11
Not provided	Ames test in <i>E coli</i> strain uvrA and <i>S</i> <i>typhimurium</i> strains TA98, TA100, TA1535, and TA1537 with and without metabolic activation by S9; plates exposed to test material in gas-sampling bags	Not mutagenic	12
0%, 35%, 50%, or 70% in 3 h exposure with and without metabolic activation; 0%, 35%, 50%, or 70% in 19 h exposure without metabolic activation; 0%, 50%, 60%, or 70% in 19 h exposure without metabolic activation	Chromosomal aberration test in human lymphocytes, with and without S9 metabolic activation (in accordance with OECD TG 473); cultures exposed to test material in gas-sampling bags	Weakly clastogenic at 60% and 70% without metabolic activation after 19 h exposure; negative with and without metabolic activation after 3 h exposure	3,11
In vivo			
0, 4875, 9750, or 19 500 ppm	Micronucleus assay in male and female Sprague-Dawley rats; animals exposed via whole body inhalation for 6 h (in accordance with OECD TG 474)	No mortality or adverse clinical signs were observed during the study; no evidence of chromosome damage or bone marrow cell toxicity	3,11

During the study, no statistically significant differences in body weights or body weight gains were observed. Clinical signs of toxicity observed included ocular/nasal discharge, wet/stained perinea, stained body/face, and/or swollen ears. These clinical signs were also observed in some control animals. Clinical chemistry effects included increased mean corpuscular volumes, increased serum bilirubin, increased hematocrits, and/or increased urobilinogen. Because there were no abnormalities in hematopoietic tissues or red blood cells or changes in serum bilirubin, there was no conclusive evidence of a hemolytic effect. A decrease in peripheral circulating eosinophils and/or monocytes was observed. A dose-dependent increase was observed in urinary fluoride concentration, but there was no evidence of fluorosis. An increase in serum creatinine and urine volume and a decrease in urine osmolality were observed in female rats. Upon study conclusion, no treatment-related differences in organ weights were observed in male rats, but significant increases in absolute and relative lung weights, absolute and relative stomach weights, relative heart weight, and relative liver weight were observed in female rats at all concentrations. The biological significance of these observations in the female rats is unknown. No treatment-related tumors were observed in male and female rats. The authors of the study concluded that

Hydrofluorocarbon 152a was not carcinogenic and did not produce life-shortening toxic effects in rats in this 2-year inhalation study.^{3,11}

Dermal Irritation and Sensitization

Dermal Irritation

No relevant published dermal irritation studies on Hydrofluorocarbon 152a were identified in a literature search for this ingredient, and no unpublished data were submitted. These studies are considered technically not feasible for gases.¹¹

Dermal Sensitization

No relevant published dermal sensitization studies on Hydrofluorocarbon 152a were identified in a literature search for this ingredient, and no unpublished data were submitted. These studies are considered technically not feasible for gases.¹¹

Ocular Irritation Studies

A hair spray containing 80% Hydrofluorocarbon 152a was considered not irritating to the eye in a bovine corneal permeability (BCOP) assay.¹³ The assay was performed based on methods described in Organization for Economic Co-Operation and Development (OECD) test guideline 437. The test material, the positive control (ethanol), and the negative control (sterile deionized water) were applied via aerosol sprays at 1 second burst from a distance of 10 cm. The mean amount of the test material sprayed on each cornea was 0.9 ± 0.19 g. Four to five corneas each were treated with each test article. The in vitro score was 2.3 for the test material while it was 48.4 for the positive control.

In the two-year inhalation study in rats described above, clinical signs of toxicity included ocular/nasal discharge.^{3,11} No further details were provided.

Clinical StudiesCase

Case Reports

Numerous case reports of adverse events from abusive inhalation of products containing Hydrofluorocarbon 152a have been described in the literature. Adverse events include death, cardiomyopathy, cardiac arrhythmia and other cardiac and respiratory effects, rhabdomyolysis, fulminant hepatitis, acute kidney injury, angioedema, frostbite, chemical burns, and even thermal burns.¹⁴⁻¹⁸

Risk Assessment

The American Industrial Hygiene Association (AIHA) 8 hour workplace environmental exposure limit for Hydrofluorocarbon 152a is 1000 ppm.¹⁹

The U.S. Environmental Protection Agency's (EPA) Integrated Risk Information System (IRIS) has estimated the reference concentration for chronic inhalation exposure (RfC) for Hydrofluorocarbon 152a to be 40 mg/m³, which was calculated based on an uncertainty factor of 300, a modifying factor of 1, and a no observed adverse effect level (NOAEL) value of 67 500 mg/m³ (25 000 ppm) based on the value obtained from the carcinogenicity study described above.²⁰

Summary

Hydrofluorocarbon 152a, commonly known as 1,1-difluoroethane, is a gas that functions as a propellant in personal care products.

According to 2017 VCRP data, Hydrofluorocarbon 152a is used in 467 formulations; the majority of uses are in leave-on hair care products. The results of the concentration of use survey conducted in 2015 by the Council indicate the highest reported maximum concentration of use to be 80% in hair sprays.

Hydrofluorocarbon 152a may be used as an aerosol propellant, a foam expansion agent, a refrigerant, and as a catalyst regenerator.

Metabolites identified in urine collected from male rats exposed via inhalation to 3000 ppm Hydrofluorocarbon 152a included fluoride ion and a trace of acetyl fluoride.

In an uptake, distribution, and elimination inhalation study of Hydrofluorocarbon 152a in human subjects exposed to 0, 200, or 1000 ppm, initial increases of Hydrofluorocarbon 152a in the blood were fast, and within 4 h post-exposure, the concentration was less than 1% of the steady-state level. Blood concentrations were below detection limits 22 h post-exposure. Total inhaled Hydrofluorocarbon 152a was approximately 20.6 and 99.6 mmol for the 200 and 1000 ppm exposures, respectively. Post-exposure decreases of the test material in exhaled air and urine were similar to that in blood. After exposure to 200 ppm and 1000 ppm Hydrofluorocarbon 152a, about 0.004% and 0.009%, respectively, of the total amount inhaled was excreted in the urine within 23 h. Fluoride excretion rate was varied; however, it was significantly higher in the first 2 urine samples after exposure to 1000 ppm exposure. No fluorine-containing metabolites could be detected in the urine, indicating biotransformation of the test material in humans was very low.

In a rat oral dose study, the LD_{Lo} for Hydrofluorocarbon 152a was reported to be greater than 1500 mg/kg (this study was considered invalid by test sponsors because an unsuitable test system was used). The reported LC_{50} of animals in a mouse inhalation study was 977 200 ppm. Cardiac arrhythmia was observed in dogs exposed for 5 min to 150 000 ppm Hydrofluorocarbon 152a in an acute inhalation study.

In a two-week inhalation study of 100 000 ppm Hydrofluorocarbon 152a, rats that received the test material 6 h/d for 5 d/wk appeared to be anesthetized, which was indicated by sleep and unresponsiveness to sound. No other adverse effects were observed. A slight increase in urinary fluoride was observed following the final exposure. In another short-term inhalation study, rats exposed to 100 000 ppm Hydrofluorocarbon 152a for 16 h/d had no clinical signs of toxicity during 2 months of exposure. Necropsy indicated no adverse changes. Mild diffuse infiltration of small and large round cells in the lung was observed during microscopic examination, which indicated mild chronic irritation.

No treatment-related clinical signs of toxicity or body weight changes were observed in the dams (rats) of a maternal and developmental toxicity study that were exposed to up to 50 000 ppm Hydrofluorocarbon 152a in air 6 h/d on gestation days 6–15 in rats. No statistically significant differences between the control group and test groups were observed in pregnancy or fetal parameters. The NOEL for maternal and developmental toxicity in rats was 50 000 ppm.

Hydrofluorocarbon 152a was not mutagenic in Ames tests at concentrations up to 75%, but it was weakly clastogenic at 60% and 70% (19 h exposure without metabolic activation) in a chromosomal aberration test in human lymphocytes. In a rat micronucleus assay, Hydrofluorocarbon 152a was not genotoxic at concentrations up to 19 500 ppm.

The authors of a two-year inhalation study of rats exposed to concentrations up 25 000 ppm Hydrofluorocarbon 152a for 6 h/d, 5 d/wk concluded that this chemical was not carcinogenic and did not produce life-shortening toxic effects.

A hair spray containing 80% Hydrofluorocarbon 152a was considered not irritating to the eye in a BCOP assay.

Numerous case reports of adverse events from abusive inhalation of products containing Hydrofluorocarbon 152a have been described in the literature. Adverse events include death, cardiomyopathy, cardiac arrhythmia and other cardiac and respiratory effects, rhabdomyolysis, fulminant hepatitis, acute kidney injury, angioedema, frostbite, chemical burns, and even thermal burns.

The AIHA 8 hour workplace environmental exposure limit for Hydrofluorocarbon 152a is 1000 ppm. The EPA's IRIS has estimated the RfC for chronic inhalation exposure for Hydrofluorocarbon 152a to be 40 mg/m³.

Dermal irritation and sensitization studies are considered technically not feasible for gases, including Hydrofluorocarbon 152a.

Discussion

Hydrofluorocarbon 152a is a largely inert gas that is rapidly volatilized and dispersed upon application and, if incidentally inhaled and absorbed into the blood stream, is quickly cleared from the body by exhalation. Significant dermal exposures to this ingredient and potential effects during normal use of cosmetic spray products are unlikely because of the chemical and physical properties of the ingredient. The available evidence indicates that Hydrofluorocarbon 152a is not metabolized in the body to any significant extent. Extensive inhalation exposure studies of Hydrofluorocarbon 152a indicate that this ingredient in cosmetic spray products will not cause adverse health effects when these products are used as intended. These studies include acute, subchronic, and chronic studies at concentrations orders of magnitude greater than the occupational exposure limit for this compound and greater still than concentrations that can reasonably be expected during cosmetic use. The Panel found the overall safety profile of this ingredient to be favorable.

The Panel noted that the European Union has issued regulations restricting the use of fluorinated gases in personal care and household products. The regulations are directed toward protection of the global environment, which falls outside of the Panel's purview of personal use safety.

Conclusion

The Expert Panel for Cosmetic Ingredient Safety concluded that Hydrofluorocarbon 152a is safe in the present practices of use and concentration described in this safety assessment.

Author's Note

Unpublished sources cited in this report are available from the Director, Cosmetic Ingredient Review, 1620 L Street, NW, Suite 1200, Washington, DC 20036, USA.

Author Contribution

Burnett, C. contributed to conception and design, contributed to acquisition, analysis, and interpretation, drafted manuscript, and critically revised manuscript; Bergfeld, W., Belsito, D., Hill, R., Klaassen, C., Liebler, D., Marks, J., Shank, R., Slaga, T., Snyder, P., and Gill, L.J. contributed to conception and design, contributed to analysis and interpretation, and critically revised manuscript; Heldreth, B. contributed to design, contributed to analysis and interpretation, and critically revised manuscript. All authors gave final approval and agree to be accountable for all aspects of work ensuring integrity and accuracy.

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References

- Nikitakis J, Lange B. International Cosmetic Ingredient Dictionary and Handbook. 16th ed.. Washington, DC: Personal Care Products Council; 2016.
- Lewis RJ. *Hawley's Condensed Chemical Dictionary*. 15th ed.. Hoboken, NJ: John Wiley & Sons, Inc.; 2007.
- Organisation for Economic Co-Operation and Development Screening Information Data Sets (OECD SIDS). *1,1-Diffuoroethane (HFC-152a): CAS No. 75-37-6.* UNEP Publications; 2006. http://webnet.oecd.org/HPV/UI/handler.axd?id=6415a8cf-4a7b-4c8e-b943-f61ed5304b0d. Date Accessed 5-17-2016.
- Food and Drug Administration (FDA). Frequency of Use of Cosmetic Ingredients. FDA Database. Washington, DC: FDA; 2017.
- Personal Care Products Council. Updated Concentration of Use by FDA Product Category: Hydrofluorocarbon 152a; 2016. Unpublished data submitted by Personal Care Products Council.
- European Union. Regulation (EU) No. 517/2014 of the European Parliament and of the Council of 16 April 2014 on fluorinated greenhouse gases and repealing Regulation (EC) No. 842/2006. 2014. http://eur-lex.europa.eu/legal-content/EN/TXT/PDF/?uri= CELEX:32014R0517&from=EN. Accessed on February 16, 2017
- European Union. Commission Implementing Regulation (EU) 2015/2068 of 17 November 2015 establishing, pursuant to Regulation (EU) No. 517/2014 of the European Parliament and of the Council, the format of labels for products and equipment containing fluorinated greenhouse gases. 2015. http://eur-lex.

europa.eu/legal-content/EN/TXT/PDF/?uri=CELEX: 32015R2068&from=EN. Accessed on February 16, 2017

- Gluckman Consulting. EU F-Gas Regulation Guidance Information Sheet 8: Aerosols. http://gluckmanconsulting.com/ wp-content/uploads/2014/12/IS-8-Aerosols.pdf. Last Updated 2014Date Accessed 2-16-2017
- Keller DA, Roe DC, Lieder PH. Fluoroacetate-mediated toxicity of fluorinated ethanes. *Fundam Appl Toxicol.* 1996;30(2): 213-219.
- Ernstgård L, Sjögren B, Dekant W, Schmidt T, Johanson G. Uptake and disposition of 1,1-difluoroethane (HFC-152a) in humans. *Toxicol Lett.* 2012;209:21-29.
- European Chemicals Agency. 1,1-Difluoroethane. http:// echa.europa.eu/. Last Updated 2016. Date Accessed 5-12-2016.
- Araki A, Noguchi T, Kato F, Matsushima T. Improved method for mutagenicity testing of gaseous compounds by using a gas sampling bag. *Mutat Res.* 1994;307(1):335-344.
- Anonymous. Summary: Bovine Corneal Permeability (BCOP) Assay with Aerosol Dosing (Hair Spray Containing 80% Hydrofluorocarbon 152a) and Optional Histology. Care Products Council; 2016. 1. Unpublished data submitted by Persona.
- Kumar S, Joginpally T, Kim D, Yadava M, Norgais K, Laird-Fick HS. Cardiomyopathy from 1,1-difluoroethane inhalation.[E-pub ahead of print]. *Cardiovasc Toxicol*. 2015.
- Kurniali PC, Henry L, Kurl R, Meharg JV. Inhalant abuse of computer cleaner manifested as angioedema. *Am J Emerg Med*. 2012;30(1):265.e3-265.e5.
- Moreno C, Beierle EA. Hydrofluoric acid burn in a child from a compressed air duster. *J Burn Care Res.* 2007;28(6):909-912.
- Avella J, Wilson JC, Lehrer M. Fatal cardiac arrhythmia after repeated exposure to 1,1-difluoroethane (DFE). *Am J Forensic Med Pathol*. 2006;27(1):58-60.
- Kuspis DA, Krenzelok EP. Oral frostbite injury from intentional abuse of a fluorinated hydrocarbo. *J Toxicol Clin Toxicol*. 1999; 37(7):873-875.
- American Industrial Hygiene Association (AIHA). WEEL Values; 2011. Fairview Park Dr Suite 777, Falls Church, VA 22042. https://www.aiha.org/get-involved/AIHAGuidelineFoundation/ WEELs/Documents/2011WEELValues.pdf.2013 ERPG/WEEL Handbook.3141. Last Updated 2013Date Accessed 5-17-2016.
- U.S. Environmental Protection Agency. Integrated Risk Information System (IRIS): 1,1-Difluoroethane; CASRN 75-37-6. https://cfpub.epa.gov/ncea/iris/iris_documents/documents/subst/ 0665_summary.pdf. Last Updated 1994Date Accessed 5-18-2016